

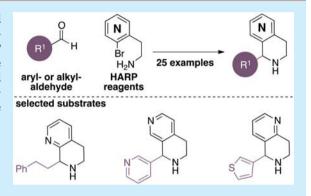
Synthesis of Tetrahydronaphthyridines from Aldehydes and HARP Reagents via Radical Pictet—Spengler Reactions

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Supporting Information

ABSTRACT: The combination of aldehydes with newly designed HARP (halogen amine radical protocol) reagents gives access to α -substituted tetrahydronaphthyridines. By using different HARP reagents, various regioisomeric structures can be prepared in a single operation. These products, which are of high value in medicinal chemistry, are formed in a predictable manner via a formal Pictet—Spengler reaction of electron-poor pyridines that would not participate in the corresponding polar reactions.



F used aromatic/saturated N-heterocycles are important scaffolds for biologically active pharmaceuticals and natural products. While many widely used examples, including tetrahydroisoquinolines and β-tetrahydrocarbinols, can be accessed by established processes such as Pictet–Spengler or Bischler–Napieralski reactions, these approaches are not well suited for the preparation of electron-deficient systems such as tetrahydronaphthyridines. The poor synthetic access to such compounds is exemplified by recent work from Amgen, where the target molecule synthesis relied on introduction of the substituents in the very first step of a long synthetic sequence. While such an approach is suitable once a lead compound is identified, it is not appealing in early stages where library syntheses and structure—activity relationship studies are the primary goal.

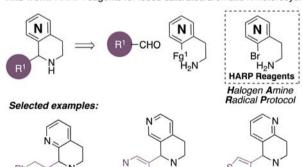
As part of a program aimed at establishing predictable, crosscoupling approaches to saturated N-heterocycles from widely available components, our group has developed SnAP (stannylamine protocol) reagents, which enables the single-step synthesis of diverse, saturated N-heterocycles from aldehydes (Scheme 1). We sought to extend this concept to the preparation of partially saturated bicyclic heterocycles, beginning with otherwise difficult to access tetrahydronaphthyridines in various regioisomeric forms. In this paper, we document our successful development of a new class of reagents, HARP (halogen amine radical protocol) reagents and conditions suitable for this cross-coupling. Although the mechanism involves radical additions to imines, the key step of the SnAP chemistry, our solution to these product classes relies on a different radical source and reaction conditions. The new approach not only allows the assembly of fused saturated/ aromatic N-heterocycle scaffolds but also delivers the expected products in a predictable manner, overcoming the inherent

Scheme 1. SnAP and HARP Reagents

Prior work: SnAP reagents for saturated N-heterocycle synthesis

$$\begin{array}{c} X \\ N \\ H \end{array} \implies \begin{array}{c} \mathbb{R}^1 - \text{CHO} \ \ \mathsf{Fg}_{1}^1 \\ \mathsf{H}_{2}^1 \mathsf{N} \end{array} \qquad \begin{array}{c} X \\ \mathsf{Bu}_3 \mathsf{Sn} \\ \mathsf{H}_2 \mathsf{N} \end{array}$$
 SnAP Reagents

This work: HARP reagents for fused saturated/aromatic N-heterocycles



Scheme 2. Synthesis of HARP Reagents

regioselectivity problems associated with the Pictet-Spengler or Bischler-Napieralski reactions.

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Scheme 3. Synthesis of 1,7-Tetrahydronaphthyridines^a

^aReaction conditions for imine formation: azido reagent (1.00 equiv), aldehyde (0.95-1.00 equiv), polymer-bound triphenylphosphine (2.0 equiv), THF, 55 °C, 12 h. Conditions for cyclization: imine (1.0 equiv), AIBN (0.2 equiv), (TMS)₃SiH (1.5-2.0 equiv), toluene, 100 °C. Yields refer to isolated amine products after purification by column chromatography. ^bAlongside product 5m, the corresponding cyclic imine was obtained in 24% yield (see the Supporting Information)

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Although aryltin reagents could serve as the source of radicals (Fg¹ in Scheme 1), we sought to identify a starting material that was both easier to prepare and avoided the use of organostannes. We reasoned that the aromatic halides should be sufficiently stable and would generate the corresponding sp²centered radical upon treatment with a radical initiator. therefore began our investigations with the preparation of 2bromopyridine reagent 3 from the corresponding aldehyde on a multigram scale (Scheme 2).5

The azide could be reduced to the amine, but we found it to be air and moisture insensitive and elected to use this reagent for imine formation. Staudinger reduction with polymer-bound

Scheme 4. 1,6- and 2,7-Tetrahydronaphthyridine Synthesis^a

^aReaction conditions for imine formation: azido reagent (1.00 equiv), aldehyde (0.95-1.00 equiv), polymer-bound triphenylphosphine (2.0 equiv), THF 55 °C, 12 h. Conditions for cyclization: imine (1.0 equiv), AIBN (0.2-1.0 equiv), (TMS)₃SiH (1.5-2.0 equiv), toluene 100 °C. Yields refer to isolated amine products after purification by column chromatography.

triphenylphosphine followed by an aza-Wittig reaction with the corresponding aldehyde cleanly gave the requisite imines, which were isolated by simple filtration from the resin.^{7d} This approach also allayed fears of intramolecular cyclization of the amine to form an indoline. Using 4-(trifluoromethyl)benzaldehyde as a substrate for reaction optimization, we tested several conditions for the radical generation and cyclization. The use of Et₃B/O₂¹⁰ or AIBN/Bu₃SnH¹¹ led to complex product mixtures, and iridium¹² or organocatalytic photoredox systems did not exhibit the desired reactivity. We were, however, pleased to find that a catalytic amount of AIBN in combination with (TMS)₃SiH¹³ provided the cyclized product, which was isolated in 84% yield after column chromatography. Encouraged by these results, we continued the investigation of the substrate scope. The imine formation and cyclization proceeded equally well with both electron-poor and electron-rich aldehydes and tolerated heterocyclic and Organic Letters Letter

ortho-substituted substrates. Products derived from aliphatic aldehydes could be isolated in acceptable yields. Surprisingly, aromatic aldehydes containing halogen atoms (-Cl and -Br) were suitable substrates, and we did not observe dehalogenation during the cyclization. We attribute this to the kinetically favored generation of the more stable 2-pyridyl radical.¹⁴

Given the key role of the 2-pyridyl halide for efficient radical generation, it was unclear if our concept could be extended to the construction of regioisomeric tetrahydronaphthyridines (Scheme 3). The necessary reagents, also as the azides, could be prepared by straightforward routes (see the Supporting Information), and imine formation using the phosphine resin proceeded equally well. As anticipated, the cyclization conditions employed for the first reagent (3) gave only trace amounts of the desired product. Lewis acidic additives such as BF₃·Et₂O¹⁵ or the use of Brønsted acidic solvents such as HFIP, ¹⁶ which is crucial for cyclizations using SnAP reagents, led to the decomposition of the imine intermediate.

By increasing the amount of AIBN, we could isolate the cyclic amines in their N-unprotected form (Scheme 4). The yields are considerably lower than for the 2-pyridyl halogen reagents, and therefore a subject for further investigation, but even at the current state of development HARP reagents provide an attractive route to these products. We attribute the lower yields and somewhat more restricted substrate scope to the reduced stability of the aryl radical, which does not benefit from stabilization from an adjacent heteroatom.¹⁷

This methodology serves as an expansion of our previous established SnAP reagent based synthesis of saturated N-heterocycles.

In summary, we have developed a new class of reagents to access α -substituted tetrahydronaphthyridines and their regioisomers via a formal Pictet—Spengler reaction in a predictable manner. The requisite reagents are air and moisture stable and can be prepared on a multigram scale in a few synthetic steps from readily available starting materials. Applications of this strategy to assemble other electron-poor heterocyclic substrates are underway in our laboratories.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.6b00523.

Experimental procedures and analytical data of all new compounds (PDF)

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The authors declare no competing financial interest.

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